

4.5 Angular momentum

Orbital angular momentum

Angular momentum operators	$\hat{\mathbf{L}} = \mathbf{r} \times \hat{\mathbf{p}}$	(4.101)	$\mathbf{L}$ angular momentum
	$\hat{L}_z = \frac{\hbar}{i} \left(x \frac{\partial}{\partial y} - y \frac{\partial}{\partial x} \right)$	(4.102)	$\mathbf{p}$ linear momentum
	$= \frac{\hbar}{i} \frac{\partial}{\partial \phi}$	(4.103)	$\mathbf{r}$ position vector
	$\hat{L}^2 = \hat{L}_x^2 + \hat{L}_y^2 + \hat{L}_z^2$	(4.104)	xyz Cartesian coordinates
	$= -\hbar^2 \left[\frac{1}{\sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) + \frac{1}{\sin^2 \theta} \frac{\partial^2}{\partial \phi^2} \right]$	(4.105)	$r\theta\phi$ spherical polar coordinates
Ladder operators	$\hat{L}_{\pm} = \hat{L}_x \pm i\hat{L}_y$	(4.106)	$\hbar$ (Planck constant)/(2 π)
	$= \hbar e^{\pm i\phi} \left(i \cot \theta \frac{\partial}{\partial \phi} \pm \frac{\partial}{\partial \theta} \right)$	(4.107)	$\hat{L}_{\pm}$ ladder operators
	$\hat{L}_{\pm} Y_l^{m_l} = \hbar [l(l+1) - m_l(m_l \pm 1)]^{1/2} Y_l^{m_l \pm 1}$	(4.108)	$Y_l^{m_l}$ spherical harmonics
Eigen-functions and eigenvalues	$\hat{L}^2 Y_l^{m_l} = l(l+1)\hbar^2 Y_l^{m_l} \quad (l \geq 0)$	(4.109)	l, m_l integers
	$\hat{L}_z Y_l^{m_l} = m_l \hbar Y_l^{m_l} \quad (m_l \leq l)$	(4.110)	
	$\hat{L}_z [\hat{L}_{\pm} Y_l^{m_l}(\theta, \phi)] = (m_l \pm 1)\hbar \hat{L}_{\pm} Y_l^{m_l}(\theta, \phi)$	(4.111)	
	$l\text{-multiplicity} = (2l+1)$	(4.112)	

Angular momentum commutation relations^a

Conservation of angular momentum ^b	$[\hat{H}, \hat{L}_z] = 0$	(4.113)	$\mathbf{L}$ angular momentum
			$\mathbf{p}$ momentum
			H Hamiltonian
			$\hat{L}_{\pm}$ ladder operators
$[\hat{L}_z, x] = i\hbar y$	(4.114)	$[\hat{L}_x, \hat{L}_y] = i\hbar \hat{L}_z$	(4.120)
$[\hat{L}_z, y] = -i\hbar x$	(4.115)	$[\hat{L}_z, \hat{L}_x] = i\hbar \hat{L}_y$	(4.121)
$[\hat{L}_z, z] = 0$	(4.116)	$[\hat{L}_y, \hat{L}_z] = i\hbar \hat{L}_x$	(4.122)
$[\hat{L}_x, \hat{p}_x] = i\hbar \hat{p}_y$	(4.117)	$[\hat{L}_+, \hat{L}_z] = -\hbar \hat{L}_+$	(4.123)
$[\hat{L}_z, \hat{p}_y] = -i\hbar \hat{p}_x$	(4.118)	$[\hat{L}_-, \hat{L}_z] = \hbar \hat{L}_-$	(4.124)
$[\hat{L}_z, \hat{p}_z] = 0$	(4.119)	$[\hat{L}_+, \hat{L}_-] = 2\hbar \hat{L}_z$	(4.125)
		$[\hat{L}^2, \hat{L}_{\pm}] = 0$	(4.126)
		$[\hat{L}^2, \hat{L}_x] = [\hat{L}^2, \hat{L}_y] = [\hat{L}^2, \hat{L}_z] = 0$	(4.127)

^aThe commutation of a and b is defined as $[a, b] = ab - ba$ (see page 26). Similar expressions hold for S and J .

^bFor motion under a central force.



Clebsch–Gordan coefficients^a

$\langle j, -m_j | l_1, -m_1; l_2, -m_2 \rangle = (-1)^{l_1+l_2-j} \langle j, m_j | l_1, m_1; l_2, m_2 \rangle$

$l_1 \times l_2$		m_j		
		j	j	$\dots$
m_1	m_2	coefficients		
m_1	m_2	$\langle j, m_j l_1, m_1; l_2, m_2 \rangle$		
$\vdots$	$\vdots$	$\vdots$		

$1 \times 1/2$		$3/2$	$+1/2$
$+1/2$	$+1/2$	1	$3/2$
$+1/2$	$-1/2$	$1/3$	$2/3$
$-1/2$	$+1/2$	$2/3$	$-1/3$

$3/2 \times 1/2$		2	$+1$
$+3/2$	$+1/2$	2	1
$+3/2$	$-1/2$	$1/4$	$3/4$
$+1/2$	$+1/2$	$3/4$	$-1/4$
$+1/2$	$-1/2$	$1/2$	$1/2$
$-1/2$	$+1/2$	$1/2$	$-1/2$

$2 \times 1/2$		$5/2$	$+3/2$
$+2$	$+1/2$	1	$5/2$
$+2$	$-1/2$	$1/5$	$4/5$
$+1$	$+1/2$	$4/5$	$-1/5$
$+1$	$-1/2$	$2/5$	$3/5$
0	$+1/2$	$3/5$	$-2/5$

1×1		2	$+1$
$+1$	1	2	1
$+1$	0	$1/2$	$1/2$
0	$+1$	$1/2$	$-1/2$
0	0	$1/6$	$1/2$
-1	$+1$	$1/6$	$-1/2$

$3/2 \times 1$		$5/2$	$+3/2$
$+3/2$	$+1$	1	$5/2$
$+3/2$	0	$2/5$	$3/5$
$+1/2$	$+1$	$3/5$	$-2/5$
$+3/2$	-1	$1/10$	$2/5$
$1/2$	0	$3/5$	$1/15$
$-1/2$	$+1$	$3/10$	$-8/15$

2×1		3	$+2$
$+2$	1	3	2
$+2$	0	$1/3$	$2/3$
$+1$	$+1$	$2/3$	$-1/3$
$+2$	-1	$1/15$	$1/3$
$+1$	0	$8/15$	$1/6$
0	$+1$	$1/15$	$-1/2$

$3/2 \times 3/2$		3	$+2$
$+3/2$	$+3/2$	1	3
$+3/2$	$1/2$	$1/2$	2
$+1/2$	$+3/2$	$1/2$	$-1/2$
$+3/2$	$-1/2$	$1/5$	$1/2$
$+1/2$	$+1/2$	$3/5$	$-2/5$
$-1/2$	$+3/2$	$1/5$	$-1/2$

$2 \times 3/2$		$7/2$	$+5/2$
$+2$	$+3/2$	1	$7/2$
$+2$	$1/2$	$3/7$	$4/7$
$+1$	$+3/2$	$4/7$	$-3/7$
$+2$	$-1/2$	$1/7$	$16/35$
$+1$	$+1/2$	$4/7$	$1/35$
0	$+3/2$	$2/7$	$-18/35$

2×2		4	$+3$
$+2$	2	4	3
$+2$	1	$1/2$	$1/2$
$+1$	$+2$	$1/2$	$-1/2$
$+2$	0	$3/14$	$1/2$
$+1$	$+1$	$4/7$	0
0	$+2$	$3/14$	$-1/2$

$2 \times 3/2$		$7/2$	$+5/2$
$+2$	$+3/2$	1	$7/2$
$+2$	$1/2$	$3/7$	$4/7$
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$2 \times 3/2$		$7/2$	$+5/2$
$+2$	$+3/2$	1	$7/2$
$+2$	$1/2$	$3/7$	$4/7$

^aOr “Wigner coefficients,” using the Condon–Shortley sign convention. Note that a square root is assumed over all coefficient digits, so that “ $-3/10$ ” corresponds to $-\sqrt{3/10}$. Also for clarity, only values of $m_j \geq 0$ are listed here. The coefficients for $m_j < 0$ can be obtained from the symmetry relation $\langle j, -m_j | l_1, -m_1; l_2, -m_2 \rangle = (-1)^{l_1+l_2-j} \langle j, m_j | l_1, m_1; l_2, m_2 \rangle$.



Angular momentum addition^a

Total angular momentum	$\mathbf{J} = \mathbf{L} + \mathbf{S}$	(4.128)	$\mathbf{J}, J$ total angular momentum
	$\hat{J}_z = \hat{L}_z + \hat{S}_z$	(4.129)	$\mathbf{L}, L$ orbital angular momentum
	$\hat{J}^2 = \hat{L}^2 + \hat{S}^2 + 2\hat{\mathbf{L}} \cdot \hat{\mathbf{S}}$	(4.130)	$\mathbf{S}, S$ spin angular momentum
	$\hat{J}_z \psi_{j,m_j} = m_j \hbar \psi_{j,m_j}$	(4.131)	ψ eigenfunctions
	$\hat{J}^2 \psi_{j,m_j} = j(j+1) \hbar^2 \psi_{j,m_j}$	(4.132)	m_j magnetic quantum number $ m_j \leq j$
	j -multiplicity $= (2l+1)(2s+1)$	(4.133)	j $(l+s) \geq j \geq l-s $
Mutually commuting sets	$\{L^2, S^2, J^2, J_z, \mathbf{L} \cdot \mathbf{S}\}$	(4.134)	{ } set of mutually commuting observables
	$\{L^2, S^2, L_z, S_z, J_z\}$	(4.135)	
Clebsch–Gordan coefficients ^b	$ j, m_j\rangle = \sum_{\substack{m_l, m_s \\ m_l + m_s = m_j}} \langle j, m_j l, m_l; s, m_s \rangle l, m_l\rangle s, m_s\rangle$	(4.136)	$ \cdot\rangle$ eigenstates $\langle \cdot \cdot \rangle$ Clebsch–Gordan coefficients

^aSumming spin and orbital angular momenta as examples, eigenstates $|s, m_s\rangle$ and $|l, m_l\rangle$.

^bOr “Wigner coefficients.” Assuming no L – S interaction.

Magnetic moments

Bohr magneton	$\mu_B = \frac{e\hbar}{2m_e}$	(4.137)	μ_B Bohr magneton $-e$ electronic charge $\hbar$ (Planck constant)/(2 π) m_e electron mass
Gyromagnetic ratio ^a	$\gamma = \frac{\text{orbital magnetic moment}}{\text{orbital angular momentum}}$	(4.138)	γ gyromagnetic ratio
Electron orbital gyromagnetic ratio	$\gamma_e = \frac{-\mu_B}{\hbar}$	(4.139)	γ_e electron gyromagnetic ratio
	$= \frac{-e}{2m_e}$	(4.140)	
Spin magnetic moment of an electron ^b	$\mu_{e,z} = -g_e \mu_B m_s$	(4.141)	$\mu_{e,z}$ z component of spin magnetic moment g_e electron g -factor (≈ 2.002) m_s spin quantum number ($\pm 1/2$)
	$= \pm g_e \gamma_e \frac{\hbar}{2}$	(4.142)	
	$= \pm \frac{g_e e \hbar}{4m_e}$	(4.143)	
Landé g -factor ^c	$\mu_J = g_J \sqrt{J(J+1)} \mu_B$	(4.144)	μ_J total magnetic moment
	$\mu_{J,z} = -g_J \mu_B m_J$	(4.145)	$\mu_{J,z}$ z component of μ_J m_J magnetic quantum number
	$g_J = 1 + \frac{J(J+1) + S(S+1) - L(L+1)}{2J(J+1)}$	(4.146)	J, L, S total, orbital, and spin quantum numbers g_J Landé g -factor

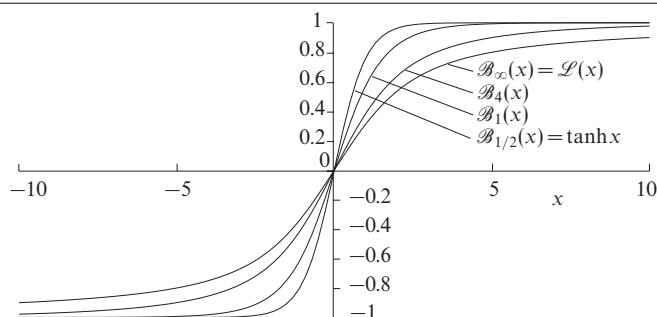
^aOr “magnetogyric ratio.”

^bThe electron g -factor equals exactly 2 in Dirac theory. The modification $g_e = 2 + \alpha/\pi + \dots$, where α is the fine structure constant, comes from quantum electrodynamics.

^cRelating the spin + orbital angular momenta of an electron to its total magnetic moment, assuming $g_e = 2$.



Quantum paramagnetism



$$\mathcal{B}_J(x) = \frac{2J+1}{2J} \coth \left[\frac{(2J+1)x}{2J} \right] - \frac{1}{2J} \coth \frac{x}{2J} \quad (4.147)$$

Brillouin
function

$$\mathcal{B}_J(x) \simeq \begin{cases} \frac{J+1}{3J} x & (x \ll 1) \\ \mathcal{L}(x) & (J \gg 1) \end{cases} \quad (4.148)$$

$$\mathcal{B}_{1/2}(x) = \tanh x \quad (4.149)$$

$\mathcal{B}_J(x)$	Brillouin function
J	total angular momentum quantum number
$\mathcal{L}(x)$	Langevin function = $\coth x - 1/x$ (see page 144)
$\langle M \rangle$	mean magnetisation
n	number density of atoms
g_J	Landé g-factor
μ_B	Bohr magneton
B	magnetic flux density
k	Boltzmann constant
T	temperature
$\langle M \rangle_{1/2}$	mean magnetisation for $J = 1/2$ (and $g_J = 2$)

Mean
magnetisation^a

$$\langle M \rangle = n \mu_B g_J \mathcal{B}_J \left(J g_J \frac{\mu_B B}{kT} \right) \quad (4.150)$$

$\langle M \rangle$ for isolated
spins ($J = 1/2$)

$$\langle M \rangle_{1/2} = n \mu_B \tanh \left(\frac{\mu_B B}{kT} \right) \quad (4.151)$$

^aOf an ensemble of atoms in thermal equilibrium at temperature T , each with total angular momentum quantum number J .

